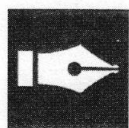


Name

Class.....

Roll No.....

Time
25 Min.Max.
Marks
17Marks
Obtained**TOPIC-2****Types of Chemical Reactions—Corrosion and Rancidity****U Q. 1.** Why are decomposing reactions called the opposite of combination reactions ?

[Board Term-I, Set (23), 2011 (NCERT)] (1)

Ans.

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U Q. 2. Why is photosynthesis considered an endothermic reaction ? [Board Term-I, Set (36), 2011] (1)

Ans.

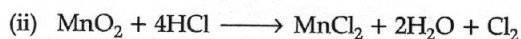
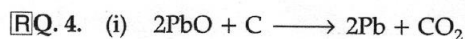
R Q. 3. What is a combination reaction ? State one example giving balanced chemical equation for the reaction.

[Board Term-I, Set (41), 2012] (2)

Ans.

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What is redox reaction ? Identify the substance oxidised and the substance reduced in the above reactions.

[Board Term-I, Set (36), 2012] (2)

Ans.

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R Q. 5. What is rancidity ? Mention any two ways by which rancidity can be prevented.

[NCT 2014, Board Term-I, Set (18), 2012] (3)

Ans.

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RQ. 6. What is meant by a precipitation reaction ? Explain by giving an example. Also give a balanced chemical equation for the reaction stating the states of the reactants and the products formed.

[Board Term-I, Set-WJ7QPA9 (NCERT), 2013] (3)

Ans.

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- RQ. 7.** (a) Define corrosion.
(b) What is corrosion of iron called ?
(c) How will you recognise the corrosion of silver ?
(d) Why corrosion of iron is a serious problem ?
(e) How can we prevent corrosion of iron ?

[Board Term-I, Set-NS9SX1D, 2016] (5)

Ans.

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